



Full communication

A one-step and label-free, electrochemical DNA detection using metal ion-mediated molecular beacon probe

Songyi Baek^{a,1}, Jun Ki Ahn^{a,1}, Byoung Yeon Won^a, Ki Soo Park^{b,*}, Hyun Gyu Park^{a,*}^a Department of Chemical and Biomolecular Engineering (BK21 + Program), KAIST, 291 Daehak-ro, Yuseong-gu, Daejeon 305-701, Republic of Korea^b Department of Biological Engineering, College of Engineering, Konkuk University, Seoul 05029, Republic of Korea

ARTICLE INFO

Keywords:

Molecular beacon
G-quadruplex
Lead ion
Electrochemistry
Biosensor

ABSTRACT

We developed a one-step and label-free, electrochemical DNA detection method using metal ion-mediated molecular beacon (MB) probe specially designed to have target-specific sequence in its loop and Pb²⁺-binding aptamer in its stem. In the absence of target DNA, MB probe, after the interaction with Pb²⁺, forms the intramolecular stem-loop hairpin structure, which limits Pb²⁺ to freely diffuse onto the electrode surface, leading to the low electrochemical signal. In contrast, the presence of target DNA that forms the hybridization complex with MB probe, breaks down the intramolecular stem-loop structure of MB probe, and releases Pb²⁺ that is freely diffused onto the electrode surface to generate the high electrochemical signal. By employing this method, the target DNA from *Chlamydia trachomatis*, one of the major pathogens causing sexually transmitted disease was successfully detected with the high selectivity.

1. Introduction

In recent years, intense interest has been aroused on the development of a simple and convenient DNA detection method that can be adopted into point-of-care-testing (POCT) and decentralized diagnostic testing [1,2]. Among many DNA detection approaches, the electrochemical methods have received the special attention due to their sensitivity, portability, operational simplicity, cost effectiveness, and low power consumption [3–5]. The representative example in this type of sensing strategy relies on the molecular beacon (MB) probe whose one end is labeled with electrochemical reporter and the other end is immobilized onto electrode surface. The presence of target DNA complementary to MB probe controls the electrochemical signal by adjusting the distance between the electrochemical reporter and the electrode surface [6–8]. In another study, Fang and co-worker suggested a similar strategy that utilizes MB probe labeled with carminic acid moieties as a key detection component [9]. Although these electrochemical methods developed to date provides the promising results, they involve the complicated and multi-step procedures including electrode modification, signaling molecule labeling, and washing [6,9–22]. These cumbersome, time-consuming procedures not only diminish the overall detection reliability of the assay, but also increase the operational costs and inconveniences, which consequently limit their practical and widespread use in POCT [23,24].

To overcome these drawbacks, metal ions have been employed as the signaling molecules for electrochemical sensors. For example, metal nanoparticles such as CdS, PbS, and ZnS conjugated to DNA probe are first captured onto the electrode through its hybridization to target DNA and the captured metal nanoparticles are then dissolved by strong acid. As a result, the released metal ions generate the highly amplified electrochemical signals [25]. These methods allow for the sensitive DNA detection, but they still entail the afore-mentioned drawbacks in the previous electrochemical methods. Therefore, there is a great need for simple, convenient, electrochemical DNA detection strategies suitable for POCT.

We herein developed a one-step and label-free, electrochemical DNA detection method using metal ion-mediated molecular beacon (MB) probe in which Pb²⁺-binding aptamer that forms G-quadruplex structure in the presence of Pb²⁺ is rationally linked to target-specific sequence. In the absence of target DNA, Pb²⁺ bound to MB probe is not freely diffused onto the electrode, generating the low electrochemical signal. In contrast, the presence of target DNA that forms the hybridization complex with MB probe, breaks down the intramolecular stem-loop structure of MB probe and releases Pb²⁺ that generates the high electrochemical signal. With this novel strategy, the target DNA from *Chlamydia trachomatis*, one of the major pathogens causing sexually transmitted disease, was analyzed with the selectivity even in the patient samples.

* Corresponding authors.

E-mail addresses: kskonkuk@gmail.com (K.S. Park), hgpark@kaist.ac.kr (H.G. Park).¹ The authors equally contributed to this work.

Table 1
Oligonucleotide sequences used in this work.

Strand name	DNA sequence (5'→3')
MB probe*	<u>GGG TAG GGT TTA AAA GGG ATT GCA GCT TGG GTT</u> <u>GGG</u>
Synthetic target DNA**	ACT ACA AGC TGC AAT CCC TTT TAA AAT AAC
Synthetic target DNA (50)**	CTC AAG CAG GAC TAC AAG CTG CAA TCC CTT TTA AAA TAA CCC CGC ACG TG
Synthetic target DNA (70)**	CGC ACG TTC TCT CAA GCA GGA CTA CAA GCT GCA ATC CCT TTT AAA ATA ACC CCG CAC GTG CTT CGA GCA A
Synthetic target DNA (90)**	CGC ACG TTC TCG CAC GTT CTC TCA AGC AGG ACT ACA AGC TGC AAT CCC TTT TAA AAT AAC CCC GCA CGT GCT TCG AGC AAC TTC GAG CAA
Synthetic target DNA (Mut-1)***	ACT ACA AGC TGC AAA CCC TTT TAA AAT AAC
Synthetic target DNA (Mut-2)***	ACT ACA AGC TGC ACA CCC TTT TAA AAT AAC
Synthetic target DNA (Mut-3)***	ACT ACA AGC TGC ACA TCC TTT TAA AAT AAC
Arbitrary DNA probe	TAC AGT CTA GGA TTC GGC GTG GGT TAA GCT
<i>C. trachomatis</i> Forward primer	AGG CGT TTG TAC TCC GTC AC
<i>C. trachomatis</i> Reverse primer	TGG TGG GGT TAA GGC AAA TCG
<i>T. vaginalis</i> Forward primer	CAC AAC ACC AAC ATA CGG CG
<i>T. vaginalis</i> Reverse primer	TGA CAG CGA GCT TAC GAA GG

*The stem region that interacts with Pb²⁺ is underlined, while the loop region that binds to the target DNA is indicated in red color.

**The region complementary to loop region in MB probe is indicated in red color.

***The mutation site of the synthetic target DNA is indicated in blue color.

2. Experimental

2.1. Materials

All the oligonucleotides were synthesized from Integrated DNA Technologies (IDT, Coralville, USA) and their sequences are listed in Table 1. The lead (II) nitrate, tris-acetate EDTA, sodium hydrogen phosphate, sodium dihydrogen phosphate, and sodium chloride were purchased from Sigma-Aldrich (St. Louis, MO, USA). All other chemicals were of analytical grade and used without further purification. Ultrapure DNase/RNase-free distilled water was purchased from Invitrogen (Carlsbad, CA, USA).

2.2. Genomic DNA isolation

Samples (30 mL) of urine specimens, collected after prostatic massage from patients infected by *Chlamydia trachomatis* and *Trichomonas vaginalis*, were centrifuged at 14,000 × g for 2 min. The centrifuged pellets were washed twice with 1 mL of PBS (0.2 M phosphate, 1.5 M sodium chloride, pH 7.4) and resuspended in 400 μL of PBS. Genomic DNA was extracted from each pellet by using an Accuprep™ Genomic DNA Extraction Kit (Bioneer, Korea) according to the manufacturer's protocol and stored at –20 °C until use.

2.3. PCR amplification

Amplification of the specific site of *Chlamydia trachomatis* gene encoding virulence proteins was performed by using the PCR method on a DNA engine-Peltier thermo cycler (Bio-Rad, Hercules, CA) in a 50 μL solution containing DNA template (1 μL), 0.25 M each primer (Table 1), 10 × PCR reaction buffer (500 mM Tris–HCl, 100 mM KCl, 50 mM (NH₄)₂SO₄, 20 mM MgCl₂), 0.2 mM dNTPs, and 1.25 U FastStart Taq DNA polymerase (Roche, Germany). PCR was programmed for 5 min at 95 °C, followed by 35 cycles of 30 s at 95 °C, 30 s at 55 °C, and 1 min at 72 °C, and 5 min at 72 °C. After the PCR amplification, the resulting PCR product (118 bp) was purified (NucleoSpin™, Macherey-Nagel, Duren, Germany) and used as the target DNA in this study. For a non-target DNA, the specific site of *Trichomonas vaginalis* gene was amplified and purified by following the same procedure used for the *Chlamydia trachomatis* except using the primer sets specific to *Trichomonas vaginalis* gene (Table 1).

2.4. Electrochemical DNA detection

The as-prepared target DNAs were heat-denatured at 95 °C for 5 min and immediately cooled [26]. The 10 μL of the target DNA solution was mixed with 40 μL of 10 mM Tris-Acetate buffer (pH 8.0) solution containing the 1.65 μM molecular beacon (MB) probe and 0.41 μM Pb²⁺. The final solution (50 μL) was incubated for 1 h at room temperature.

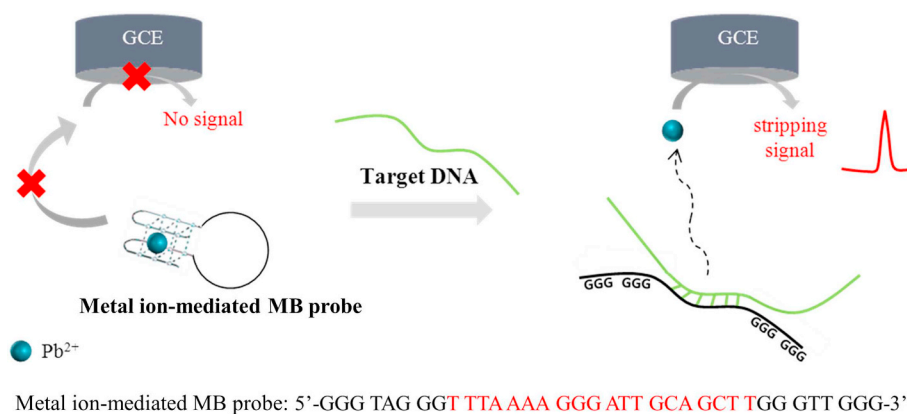


Fig. 1. Schematic illustration of a one-step and label-free, electrochemical DNA detection using metal ion-mediated molecular beacon probe.

Then, square wave anodic stripping voltammetry (SWASV) was performed to measure the electrochemical signal of Pb^{2+} by using a GAMRY Reference 600 (Warminster, PA) with screen-printed carbon electrodes consisting of a carbon working (4 mm diameter), carbon auxiliary and silver pseudoreference electrodes (DropSens, Oviedo-Asurias, Spain). SWASV was measured under the following conditions: initial potential = -0.8 V , final potential = -0.2 V , pulse size potential = 25 mV , step size = 0.05 V , frequency = 25 Hz , accumulation time = 120 s , and equilibrium time = 10 s [23,24]. For chronoamperometric studies, the cell was left at $+300\text{ mV}$ vs. Ag/AgCl reference for 20 s to reach the equilibrium state and then the current transient was measured for Pb^{2+} , Pb^{2+} with MB probe, Pb^{2+} with arbitrary DNA probe (Table 1), Pb^{2+} with MB probe and target DNA at different concentrations (100 pM and 10 nM) at -1 V (see Fig. S1 for details). The electrochemical signal differences for each case were assessed using a two-tailed t -test or one way analysis of variance (ANOVA) of the GraphPad Prism software, and P value smaller than 0.05 was considered significant.

3. Results and discussion

Fig. 1 illustrates a one-pot, electrochemical DNA detection method which utilizes metal ion-mediated molecular beacon (MB) probe rationally designed to contain target-specific sequence and Pb^{2+} -binding aptamer. In the absence of target DNA, Pb^{2+} -binding aptamer in MB probe interacts with Pb^{2+} to form G-quadruplex structure and thus Pb^{2+} bound to MB probe does not freely move onto the electrode, leading to the low electrochemical signal. On the other hand, the presence of target DNA that forms the hybridization complex with MB probe breaks down the intramolecular stem-loop structure of MB probe, and releases Pb^{2+} that is freely diffused onto the electrode surface to generate the high electrochemical signal. Based on the strong electrochemical signal of Pb^{2+} , the target DNA can be simply detected in a one-step without any modification and washing steps.

First, the ratio between MB probe and Pb^{2+} that is essential for the efficient DNA detection was optimized by measuring the electrochemical signal at ca. -0.48 V , the anodic peak potential of Pb^{2+} . As shown in Fig. S2, the electrochemical signal decreased with increasing ratio of MB probe to Pb^{2+} and became saturated at the ratio over 4, which indicates that most of Pb^{2+} were captured by MB probe, limiting its access to the electrode. Thus, the optimal ratio of 4 was selected for further experiments.

Next, the detection feasibility of the developed method was verified with the synthetic target DNA (30-mer) specific to *Chlamydia trachomatis* (*C. trachomatis*). As envisioned in the schematic illustration (Fig. 1), the low electrochemical signal was obtained in the absence of target DNA, which is attributed to Pb^{2+} bound to the stem region in MB probe. In contrast, the presence of target DNA breaks down the

intramolecular stem-loop structure of MB probe and releases Pb^{2+} that freely approaches the electrode, which was manifested by the high electrochemical signal (Fig. 2(a)). In addition, the peak current differences (Δi) increased with increasing concentrations of target DNA (*C. trachomatis*) up to 100 nM , over which it reached a plateau. An excellent linear relationship ($R^2 = 0.996$) existed in the range from 32 pM to 100 nM (Fig. 2(b)). For confirmation of performance on our system, in addition, we examined the electrochemical signaling effect on the length of target DNA (Fig. S3) and buffer condition (Fig. S4) and biological interfering agent such as nucleic acid and proteins (Fig. S5).

We also investigated the ability of our system to detect DNA mutations. The results in Fig. 3 show that in the presence of target DNA containing three base pair mutations (Mut-3), the electrochemical signal from Pb^{2+} was significantly decreased as compared to the one in the presence of complementary target DNA. In addition, the presence of target DNA with one or two base pair mutations (Mut-1 and 2) also decreases the electrochemical signal ($P < 0.0139$, unpaired two-tailed t -test). In addition, the effect of the non-complementary DNA on the hybridization of the MB probe with the target DNA was examined by measuring the electrochemical signal from Pb^{2+} . As shown in the Fig. S6, the high electrochemical signals were observed, regardless of the presence of non-complementary target DNA ($P = 0.3221$, unpaired two-tailed t -test), which clearly confirms that non-complementary target DNA does not interfere with the hybridization of the molecular beacon probe with the target DNA.

To further support the detection feasibility, the chronoamperometric analyses of Pb^{2+} , Pb^{2+} with MB probe, and Pb^{2+} with arbitrary DNA probe (Table 1), and Pb^{2+} with MB probe and target DNA (100 pM and 10 nM) were performed. The results in Fig. 4(a) show that the current decreases as the time increases, which were replotted as a function of time $^{-1/2}$ to compare the slopes of linear trend line for Pb^{2+} , Pb^{2+} with MB probe, Pb^{2+} with arbitrary DNA probe, and Pb^{2+} with MB probe and target DNA (100 pM and 10 nM). According to Cottrell Eq. (1), where i is current, n is the number of electrons transferred per Pb^{2+} , F is the Faraday constant, A is the electrode surface area, C is the total concentration of Pb^{2+} , D is the diffusion coefficient of Pb^{2+} , and t is time for which potential is applied (Fig. 4(b)) [27,28], the slope of the plot in Fig. 4(b) is only proportional to D . The diffusion coefficients for Pb^{2+} with MB probe and Pb^{2+} with arbitrary DNA probe, and Pb^{2+} with MB probe and target DNA (100 pM and 10 nM) were found to be 0.23 , 0.92 , 0.58 and 0.78 relative to that for free Pb^{2+} , respectively, which corresponds to 460.1 , 1820.5 , 635.1 , and 1369.2 nA , respectively, in SWASV. This result indicates that Pb^{2+} is selectively captured by Pb^{2+} -binding aptamer in MB probe, leading to a significant reduction of diffusion coefficient. These observations clearly confirm that the developed strategy, which relies on metal ion-mediated MB probe, can be utilized for the one-step, label-free, and electrochemical detection of target DNA.

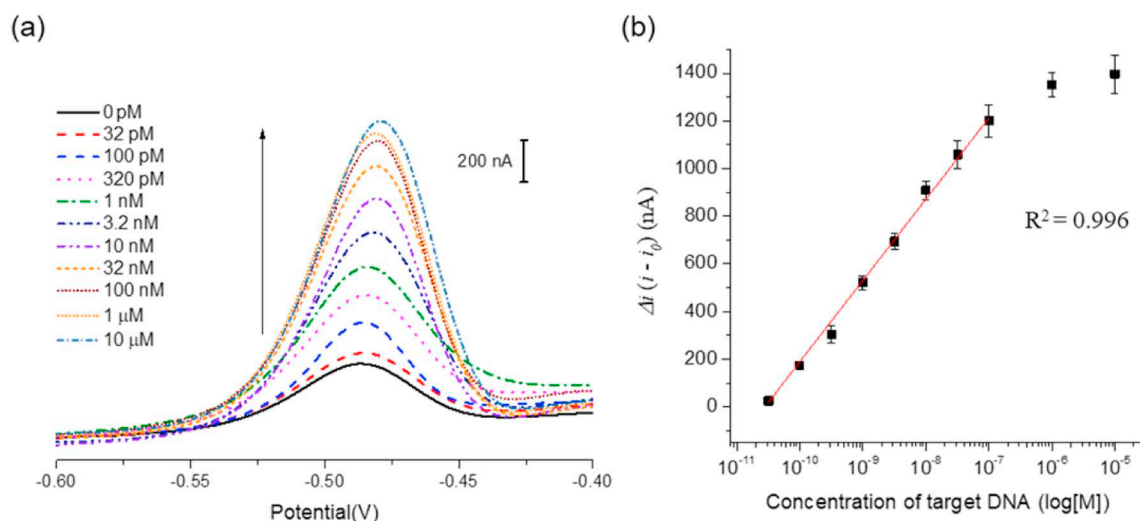


Fig. 2. Quantitative analysis of synthetic target DNA specific to *C. trachomatis*. (a) Square wave anodic stripping voltammogram (SWASV) in the presence of synthetic target DNA at varying concentrations (0, 32 pM, 100 pM, 320 pM, 1 nM, 3.2 nM, 10 nM, 32 nM, 100 nM, 1 μM, and 10 μM from bottom to top). (b) A calibration curve of peak current change (Δi) vs. log target DNA concentration. The peak current change (Δi) is defined as $i - i_0$, where i_0 and i are the electrochemical signal from Pb^{2+} bound to MB probe in the absence and presence of target DNA, respectively.

$$i = \frac{nFACD^{1/2}}{\pi^{1/2}t^{1/2}}$$

$$\left(\frac{\pi^{1/2}}{nFAC}\right) \frac{i}{t^{-1/2}} = D^{1/2} \quad (1)$$

Finally, the experiments to demonstrate the diagnostic capability of the developed method were carried out using the patient samples infected by *C. trachomatis* and *Trichomonas vaginalis* (*T. vaginalis*), major pathogens causing sexually transmitted diseases.

As shown in Fig. 5, the peak current differences (Δi) increased as the concentration of target *C. trachomatis* DNA increased. An excellent linear relationship ($R^2 = 0.998$) existed in the range from 32 pM to 100 nM and the detection limit ($3\sigma/\text{slope}$) was determined to be 69.5 pM, a value that is comparable to those from other electrochemical strategies (Table S1) [6,9–22]. Importantly, no significant current difference was observed when non-complementary DNA derived from *T. vaginalis* was tested, confirming the high specificity of the developed strategy. Overall, these results support that this method can be used for the reliable analysis of target DNA from clinical samples, verifying its practical applicability.

4. Conclusion

In summary, we have developed a one-step and label-free, electrochemical DNA detection method that utilizes metal ion-mediated MB probe as a key component. By employing this strategy, the target DNAs derived from the patient samples infected with *C. trachomatis* and *T. vaginalis* were successfully analyzed with the high selectivity. In addition, the diffusion coefficients of Pb^{2+} in different situations were compared to clearly support the proposed working principle. Importantly, this strategy does not involve multiple washing steps and expensive modifications required in the previous approaches, ensuring its widespread application in POCT. In addition, it has a huge potential for the multiplexed DNA detection and the detection of other biomolecules including small molecules, proteins, and cells, which could be accomplished by simply replacing Pb^{2+} -binding aptamers with different metal ion-binding aptamers such as mercury, silver, and copper etc. or substituting the target-specific sequence with other biomolecule-binding aptamers. Finally, we believe that this approach will pave the way for the development of new types of electrochemical sensors.

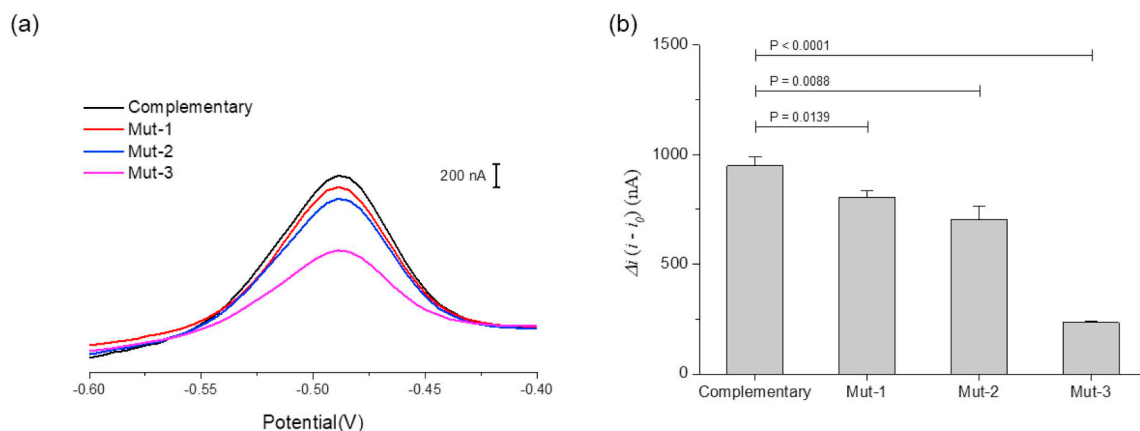


Fig. 3. DNA mutation detection feasibility. (a) Square wave anodic stripping voltammogram (SWASV) and (b) peak current change (Δi) in the presence of complementary target DNA and target DNA containing one, two, or three base mutations at the concentration of 10 nM (Table 1). The length of all DNA used in this experiment 30 bp. The peak current change (Δi) is defined as $i - i_0$, where i_0 and i are the electrochemical signal from Pb^{2+} bound to MB probe in the absence and presence of target DNA, respectively.

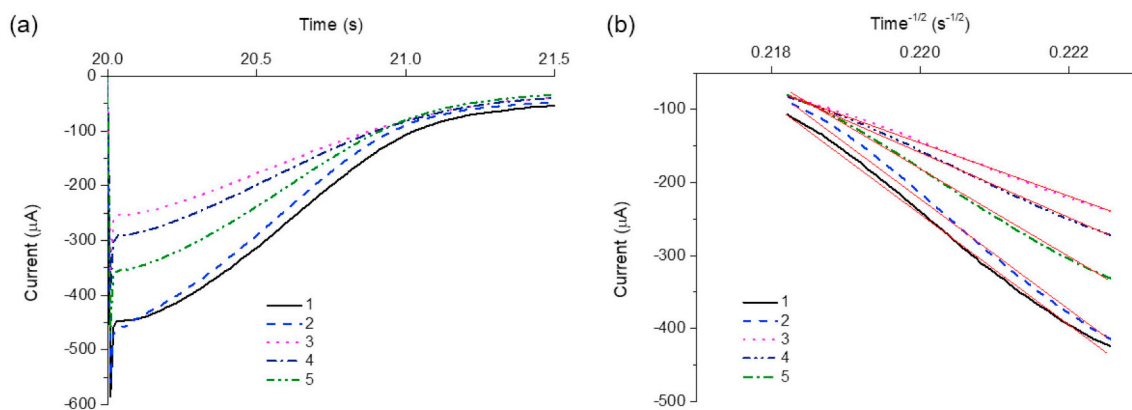


Fig. 4. Chronoamperometric analysis of free Pb^{2+} (1), Pb^{2+} with arbitrary DNA probe (2), Pb^{2+} with MB probe (3), Pb^{2+} with MB probe and target DNA at the concentration of 100 pM (4) and 10 nM (5) to compare their diffusion coefficients. The concentration of Pb^{2+} , arbitrary DNA, and MB probe was 0.33 μM , 1.32 μM , and 1.32 μM , respectively. (a) Current-time response curve (b) Cottrell plot of current vs. $\text{time}^{-1/2}$. The red line indicates the linear trend line for each Cottrell plot. For chronoamperometric studies, the applied potential to the working electrode was +300 mV vs. Ag/AgCl reference electrode. (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article.)

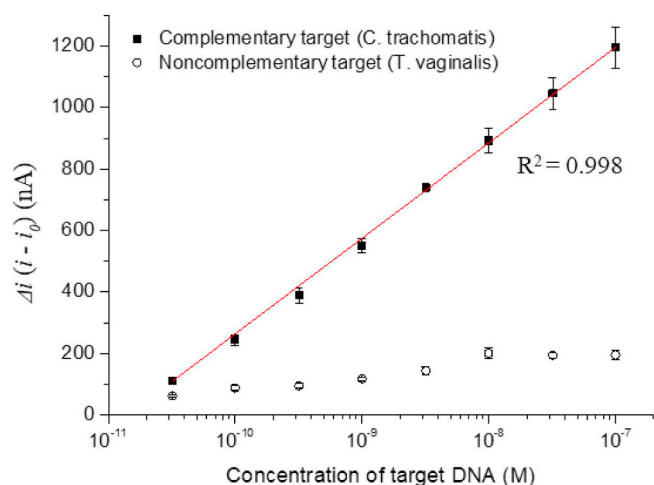


Fig. 5. Clinical application for the analysis of patient samples. The peak current change (Δi) was plotted as the log concentration of target DNA (*C. trachomatis*) and non-target DNA (*T. vaginalis*) derived from patient samples. The peak current change (Δi) is defined as $i - i_0$, where i_0 and i are the electrochemical signal from Pb^{2+} bound to MB probe in the absence and presence of target DNA, respectively.

Acknowledgements

Financial support was provided by Center for BioNano Health-Guard funded by the Ministry of Science and ICT (MSIT) of Korea as Global Frontier Project (Grant H-GUARD_2013M3A6B2078964) and Basic Science Research Program through the National Research Foundation (NRF) funded by the Ministry of Education (No. 2015R1A2A1A01005393).

Appendix A. Supplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.elecom.2019.01.023>.

References

- [1] W.G. Lee, Y.-G. Kim, B.G. Chung, U. Demirci, A. Khademhosseini, Nano/microfluidics for diagnosis of infectious diseases in developing countries, *Adv. Drug Deliv. Rev.* 62 (2010) 449–457.
- [2] B.Y. Won, H.G. Park, A touchscreen as a biomolecule detection platform, *Angew. Chem. Int. Ed.* 124 (2012) 772–775.

- [3] W.G. Kuhr, Electrochemical DNA analysis comes of age, *Nat. Biotechnol.* 18 (2000) 1042–1044.
- [4] I. Willner, Biomaterials for sensors, fuel cells, and circuitry, *Science* 298 (2002) 2407–2408.
- [5] J. Fritz, E.B. Cooper, S. Gaudet, P.K. Sorger, S.R. Manalis, Electronic detection of DNA by its intrinsic molecular charge, *Proc. Natl. Acad. Sci. U. S. A.* 99 (2002) 14142–14146.
- [6] C. Fan, K.W. Plaxco, A.J. Heeger, Electrochemical interrogation of conformational changes as a reagentless method for the sequence-specific detection of DNA, *Proc. Natl. Acad. Sci. U. S. A.* 100 (2003) 9134–9137.
- [7] F. Ricci, G. Adornetto, D. Moscone, K.W. Plaxco, G. Palleschi, Quantitative, reagentless, single-step electrochemical detection of anti-DNA antibodies directly in blood serum, *Chem. Commun.* 46 (2010) 1742–1744.
- [8] Y. Xiao, A.A. Lubin, B.R. Baker, K.W. Plaxco, A.J. Heeger, Single-step electronic detection of femtomolar DNA by target-induced strand displacement in an electrode-bound duplex, *Proc. Natl. Acad. Sci. U. S. A.* 103 (2006) 16677–16680.
- [9] J. Wu, C. Huang, G. Cheng, F. Zhang, P. He, Y. Fang, Electrochemically active-inactive switching molecular beacon for direct detection of DNA in homogenous solution, *Electrochem. Commun.* 11 (2009) 177–180.
- [10] K. Hsieh, Y. Xiao, H. Tom Soh, Electrochemical DNA detection via exonuclease and targetcatalyzed transformation of surface-bound probes, *Langmuir* 26 (2010) 10392–10396.
- [11] C. Liu, D. Jiang, G. Xiang, L. Liu, F. Liu, X. Pu, An electrochemical DNA biosensor for the detection of *Mycobacterium tuberculosis*, based on signal amplification of graphene and a gold nanoparticle–polyaniline nanocomposite, *Analyst* 139 (2014) 5460–5465.
- [12] C. Wang, H. Zhou, W. Zhu, H. Li, J. Jiang, G. Shen, R. Yu, Ultrasensitive electrochemical DNA detection based on dual amplification of circular strand-displacement polymerase reaction and hybridization chain reaction, *Biosens. Bioelectron.* 47 (2013) 324–328.
- [13] Y. Qian, T. Fan, P. Wang, X. Zhang, J. Luo, F. Zhou, Y. Yao, X. Liao, Y. Li, F. Gao, A novel label-free homogeneous electrochemical immunosensor based on proximity hybridization-triggered isothermal exponential amplification induced G-quadruplex formation, *Sensors Actuators B Chem.* 248 (2017) 187–194.
- [14] F. Zhou, Y. Yao, J. Luo, X. Zhang, Y. Zhang, D. Yin, F. Gao, P. Wang, Proximity hybridization-regulated catalytic DNA hairpin assembly for electrochemical immunoassay based on in situ DNA template-synthesized Pd nanoparticles, *Anal. Chim. Acta* 969 (2017) 8–17.
- [15] H. Wang, Y. Zhang, Y. Wang, H. Ma, B. Du, Q. Wei, Facile synthesis of cuprous oxide nanowires decorated graphene oxide nanosheets nanocomposites and its application in label-free electrochemical immunosensor, *Biosens. Bioelectron.* 87 (2017) 745–751.
- [16] L. Cao, C. Fang, R. Zeng, X. Zhao, F. Zhao, Y. Jiang, Z. Chen, A disposable paper-based microfluidic immunosensor based on reduced graphene oxide-tetraethylene pentamine/Au nanocomposite decorated carbon screen-printed electrodes, *Sensors Actuators B Chem.* 252 (2017) 44–54.
- [17] Z. Chen, Y. Liu, C. Xin, J. Zhao, S. Liu, A cascade autocatalytic strand displacement amplification and hybridization chain reaction event for label-free and ultrasensitive electrochemical nucleic acid biosensing, *Biosens. Bioelectron.* 113 (2018) 1–8.
- [18] W. Wang, T. Bao, X. Zeng, H. Xiong, W. Wen, X. Zhang, S. Wang, Ultrasensitive electrochemical DNA biosensor based on functionalized gold clusters/graphene nanohybrids coupling with exonuclease III-aided cascade target recycling, *Biosens. Bioelectron.* 91 (2017) 183–189.
- [19] S. Xu, Y. Zhang, K. Dong, J. Wen, C. Zheng, S. Zhao, Electrochemical DNA biosensor based on graphene oxide-chitosan hybrid nanocomposites for detection of *Escherichia coli* O157: H7, *Int. J. Electrochem. Sci.* 12 (2017) 3443–3458.
- [20] W. Yaqiong, H. Sauriat-Dorizon, H. Korri-Yousoufi, Direct electrochemical DNA biosensor based on reduced graphene oxide and metalloporphyrin nanocomposite,

- Sensors Actuators B Chem. 251 (2017) 40–48.
- [21] M. Cui, Y. Wang, H. Wang, Y. Wu, X. Luo, A label-free electrochemical DNA biosensor for breast cancer marker BRCA1 based on self-assembled antifouling peptide monolayer, *Sensors Actuators B Chem.* 244 (2017) 742–749.
- [22] F. Xuan, X. Luo, I.-M. Hsing, Ultrasensitive solution-phase electrochemical molecular beacon-based DNA detection with signal amplification by exonuclease III-assisted target recycling, *Anal. Chem.* 84 (2012) 5216–5220.
- [23] S. Baek, B.Y. Won, K.S. Park, H.G. Park, An electrochemical one-step system for assaying methyltransferase activity based on transport of a quantum dot signaling tracer, *Biosens. Bioelectron.* 49 (2013) 542–546.
- [24] B.Y. Won, S. Shin, R. Fu, S.C. Shin, D.-Y. Cho, H.G. Park, A one-step electrochemical method for DNA detection that utilizes a peroxidase-mimicking DNzyme amplified through PCR of target DNA, *Biosens. Bioelectron.* 30 (2011) 73–77.
- [25] J. Wang, G. Liu, A. Merkoçi, Electrochemical coding technology for simultaneous detection of multiple DNA targets, *J. Am. Chem. Soc.* 125 (2003) 3214–3215.
- [26] S.H. Hwang, W.Y. Kwon, H. Eun, S. Jeong, J.S. Park, K.J. Kim, H.J. Kim, S.H. Lee, K. Park, J.J. Yoon, Y.H. Yang, K.S. Park, The use of a 2-aminopurine-containing, split G-quadruplex for sequence-specific DNA detection, *Artif. Cells Nanomed. Biotechnol.* (2018) 1–6.
- [27] H. Heli, M. Hajizadeh, A. Jabbari, A. Moosavi-Movahedi, Copper nanoparticles-modified carbon paste transducer as a biosensor for determination of acetylcholine, *Biosens. Bioelectron.* 24 (2009) 2328–2333.
- [28] R. Sadeghi, H. Karimi-Maleh, A. Bahari, M. Taghavi, A novel biosensor based on ZnO nanoparticle/1, 3-dipropylimidazolium bromide ionic liquid-modified carbon paste electrode for square-wave voltammetric determination of epinephrine, *Phys. Chem. Liq.* 51 (2013) 704–714.